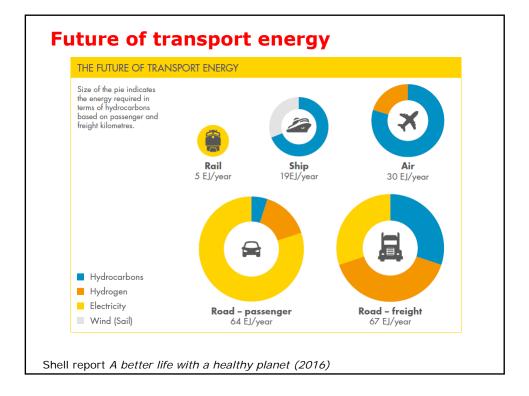
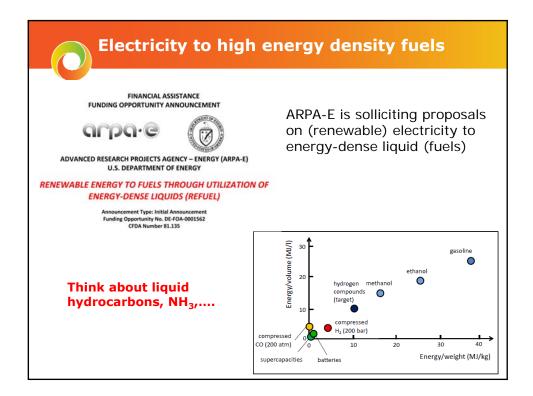


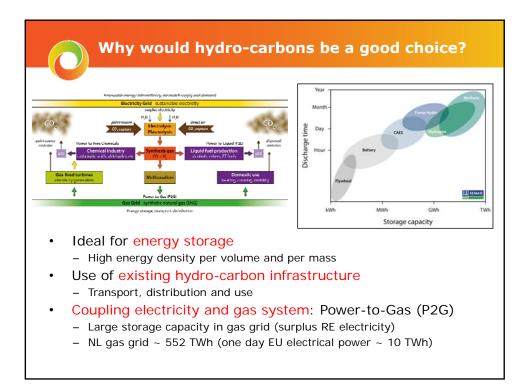
Many references to abundancy of renewable electricity and surplus electrons......

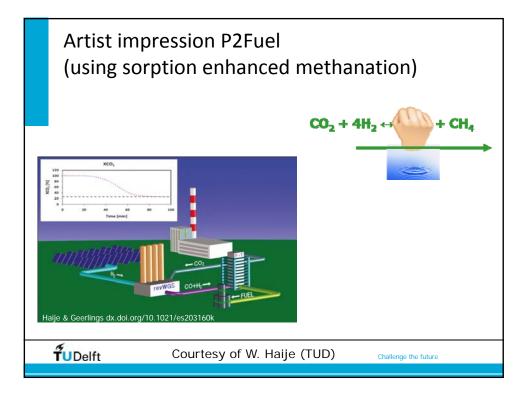
"However, the real milestone is reached when an offshore hydrogen electrolysis system is built utilising the growing surplus **electrons** from those wind farms...."

http://www.nera.nl/wp-content/uploads/2016/08/Dutch-National-Research-Agenda-Route-Energy-Transition.pdf

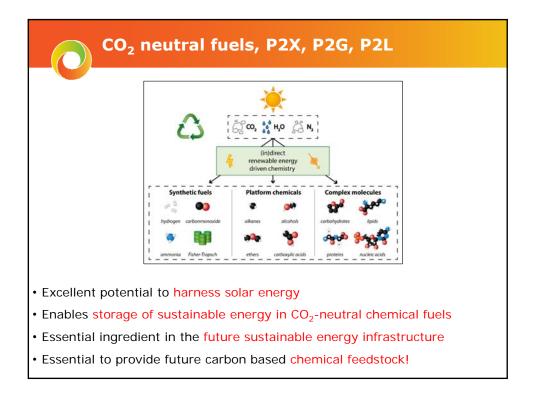


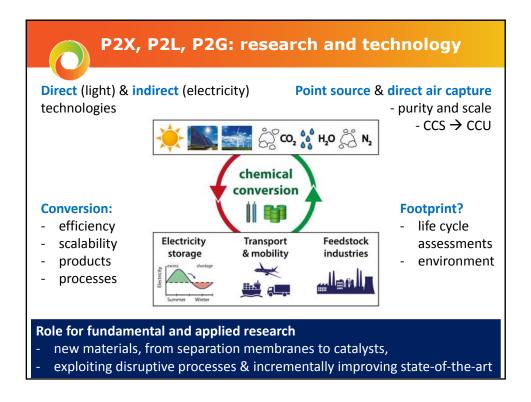


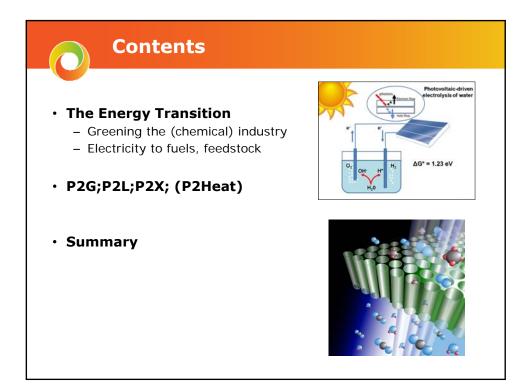


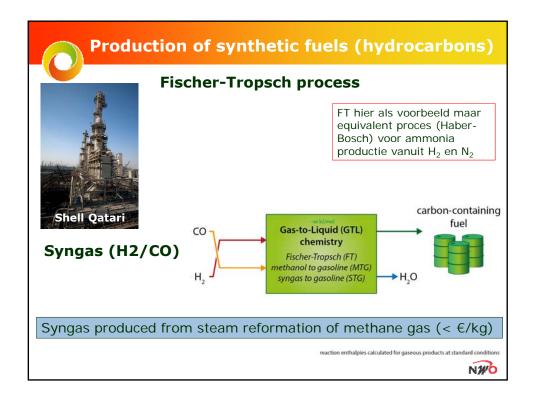


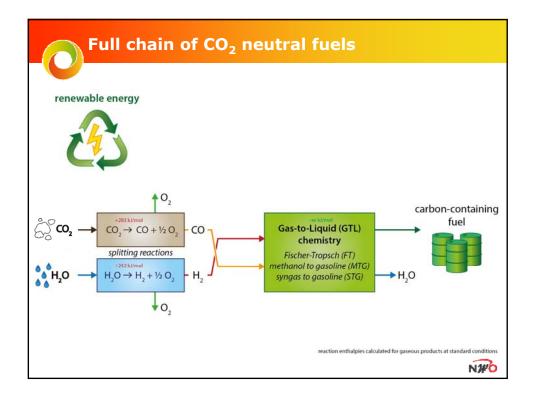
mmonia	Ammonia NH_3 for large scale energy storage			
Amn	Production	 based on abundant elements N, H, and cheap catalysts: scalable. can be produced using abundant renewable electricity and CO₂ free. 	Air H ₂ O	
	Storage	 > energy density NH₃: 22.5 MJ/kg (HHV) > liquid at 10 Bar, 20 °C > current containers can contain 60000 ton NH₃~ 375 GWh. 	Mar TWh	
	Use	 potentially clean use in fuel cell, combustion engine, gas turbine. No CO₂. fertilizer industry: CO₂ neutral fertilizers 	X15 NH ₃ jet engine Toyota GT86 R	
	Acceptance	 > poisonous, but 100⁺ yr industrial know-how > current NH₃ production costs >1.5% of world energy use >6 EJ/yr > Bio degradable (fertilizer) 	NH ₃ ship C-Job Naval Archite	
	TUDelft Courtesy of F. Mulder (TUD)			

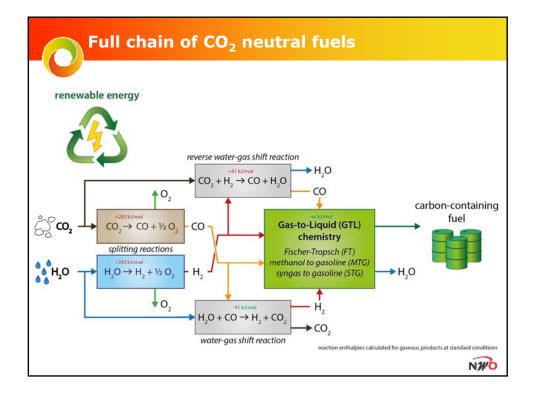


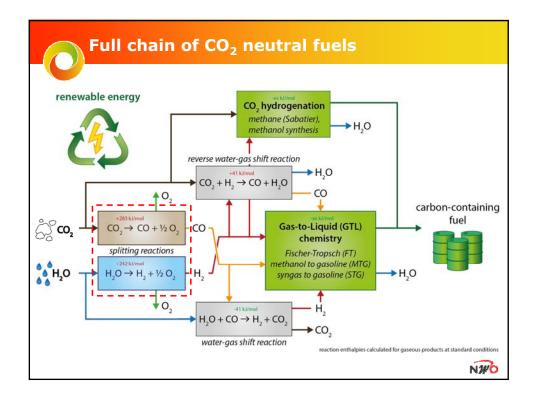


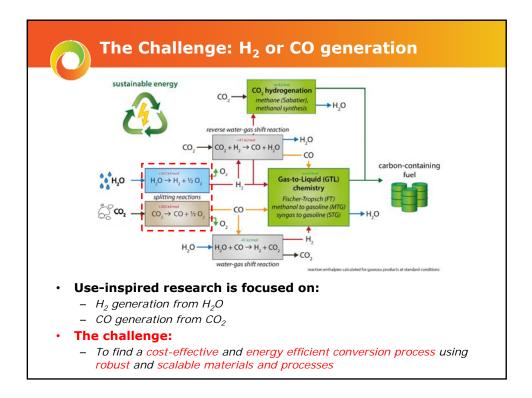


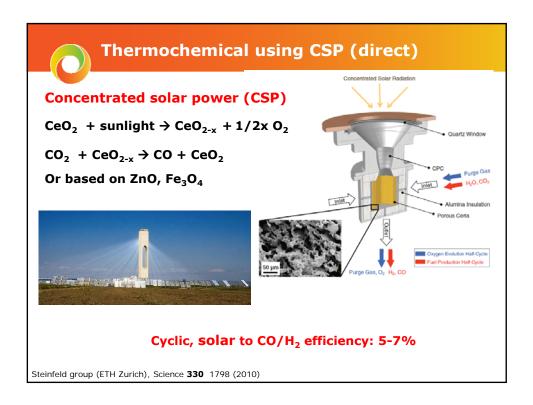


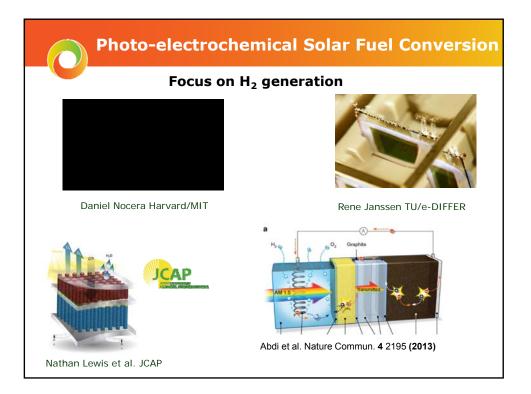


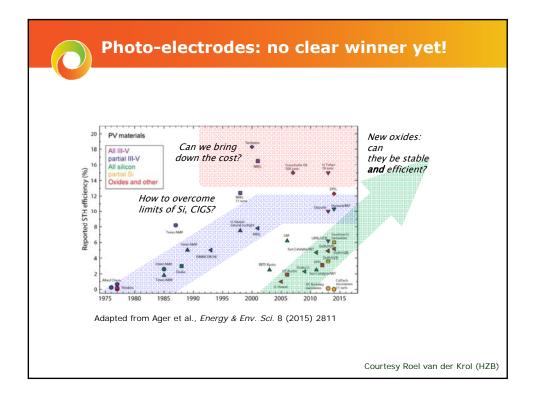


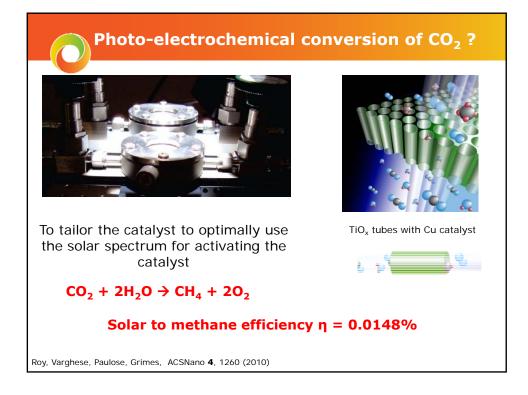


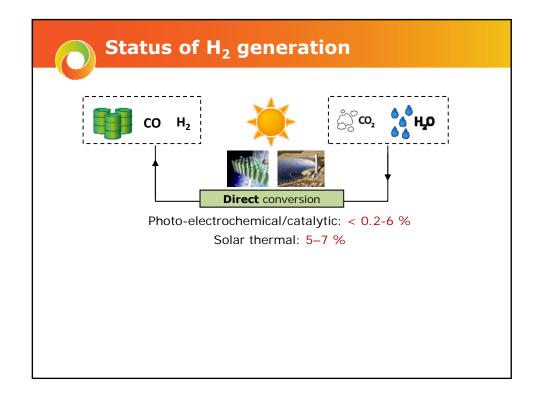


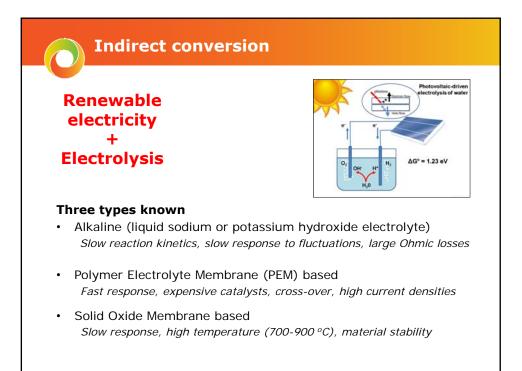


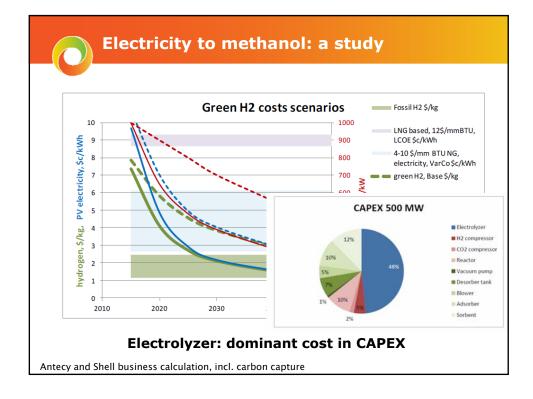


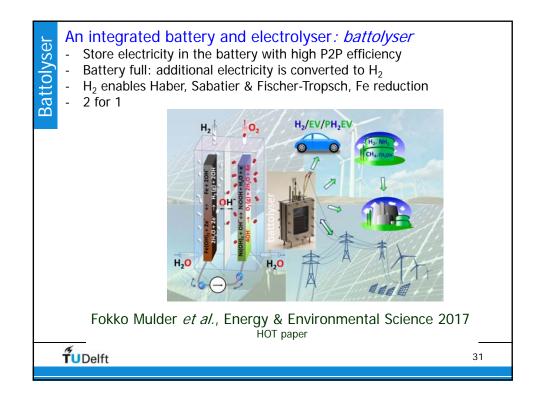












Ser	Basic reactions:
Battolyser	Negative electrode: Battery: $Fe(OH)_2 + 2e^- \leftrightarrow Fe + 2OH^-$ Electrolysis: $2H_2O + 2e^- \rightarrow H_2(g) + 2OH^-$ Positive electrode: Battery: $Ni(OH)_2 + OH^- \leftrightarrow NiOOH + H_2O + e^-$ Electrolysis: $4OH^- \rightarrow O_2(g) + 2H_2O + 4e^-$
ŤU Delft	32

